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(54) **FIELD ASYMMETRIC ION MOBILITY
SPECTROMETRY SYSTEM**

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250/297, 397

See application file for complete search history.

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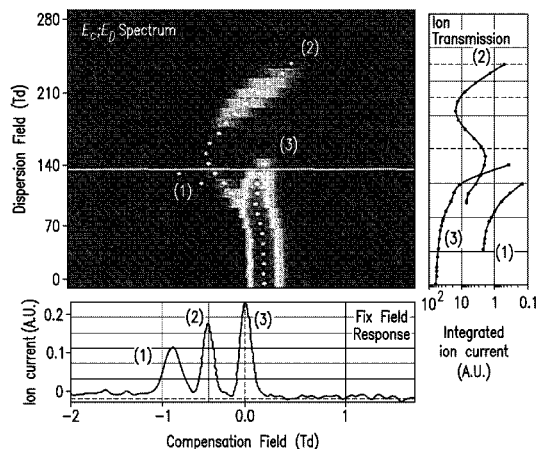
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(57) **ABSTRACT**

An apparatus, system and method for detecting, identifying, classifying and/or quantifying chemical species in a gas flow using a micro-fabricated ion filter coupled to a system adapted to apply drive signals to the ion filter. Coupled to the ion filter is a system adapted to measure the output of the ion filter, which in turn is coupled to a system adapted to extract numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow.

14 Claims, 8 Drawing Sheets



US 9,366,649 B2

Page 2

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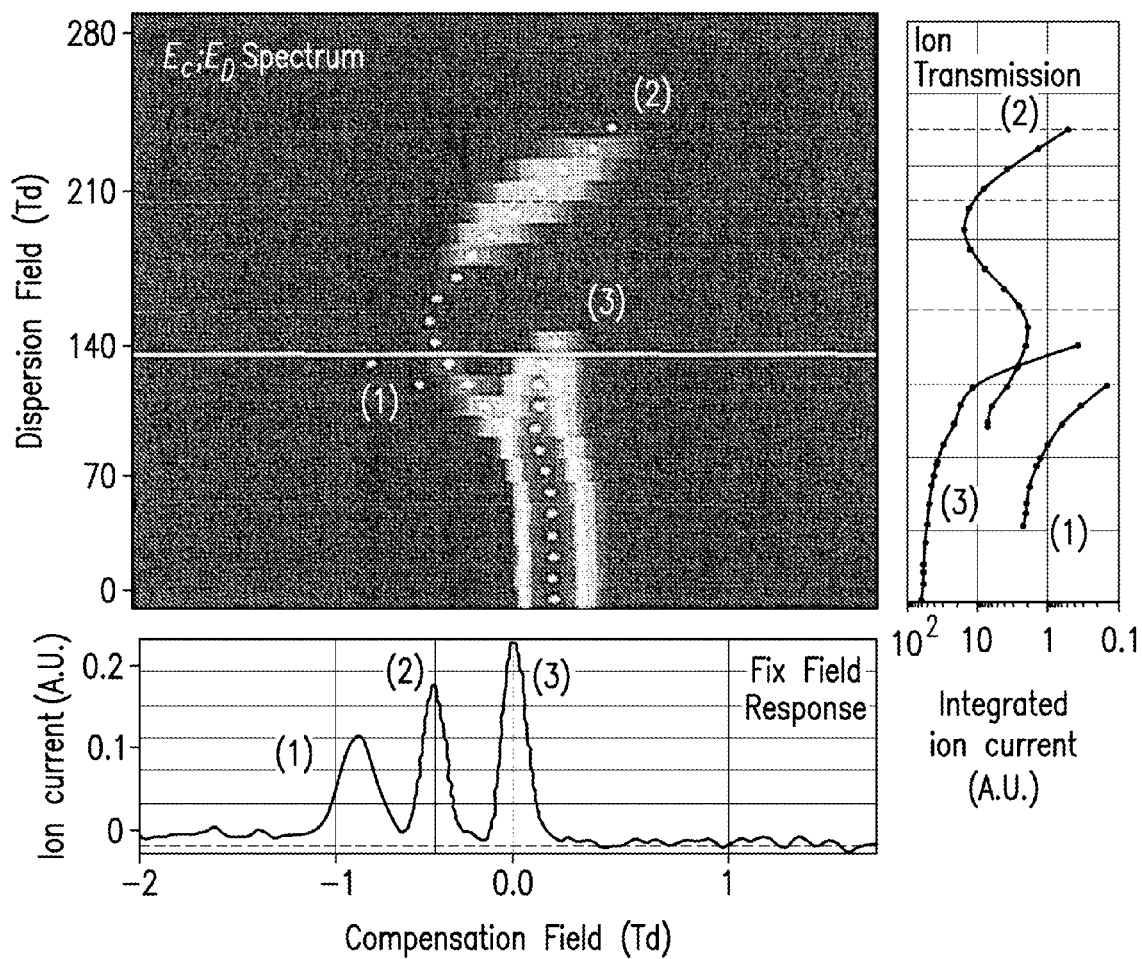


FIG. 1A

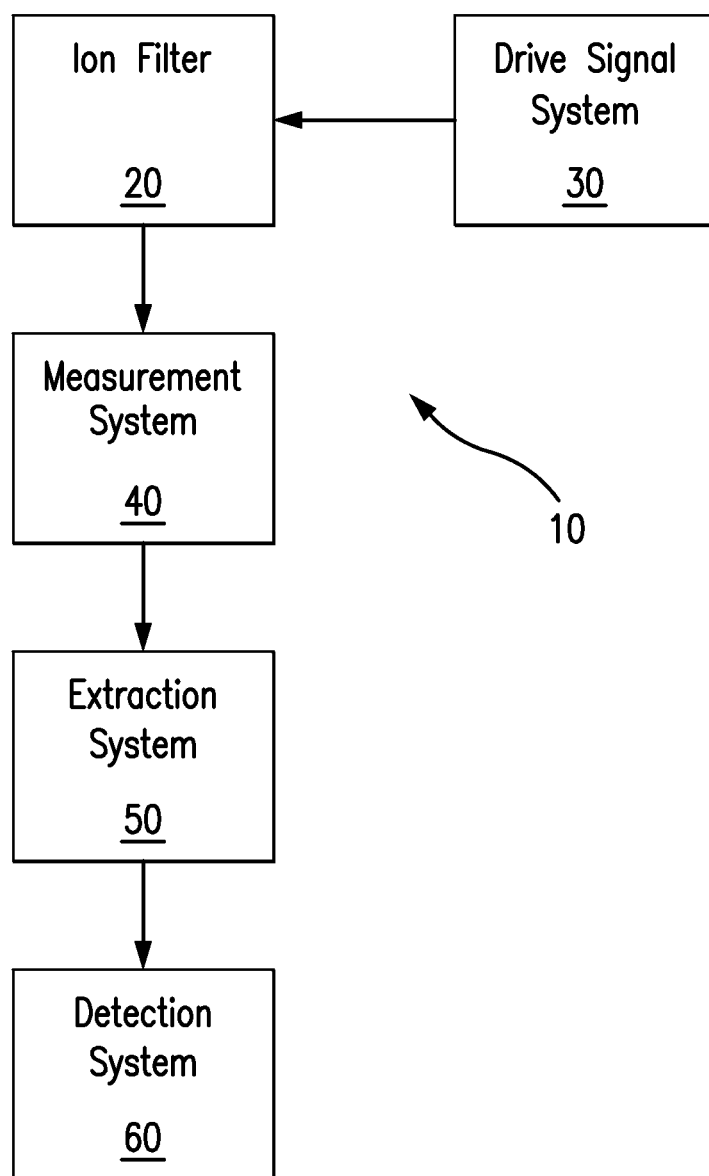


FIG. 1B

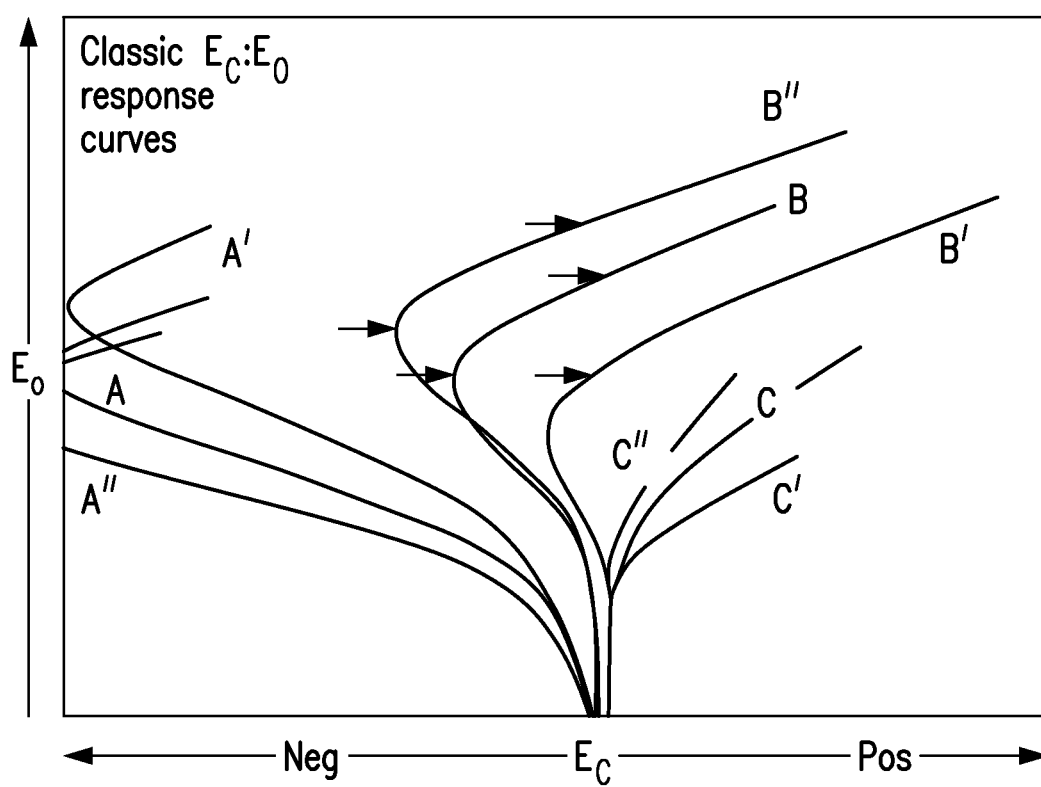


FIG. 2

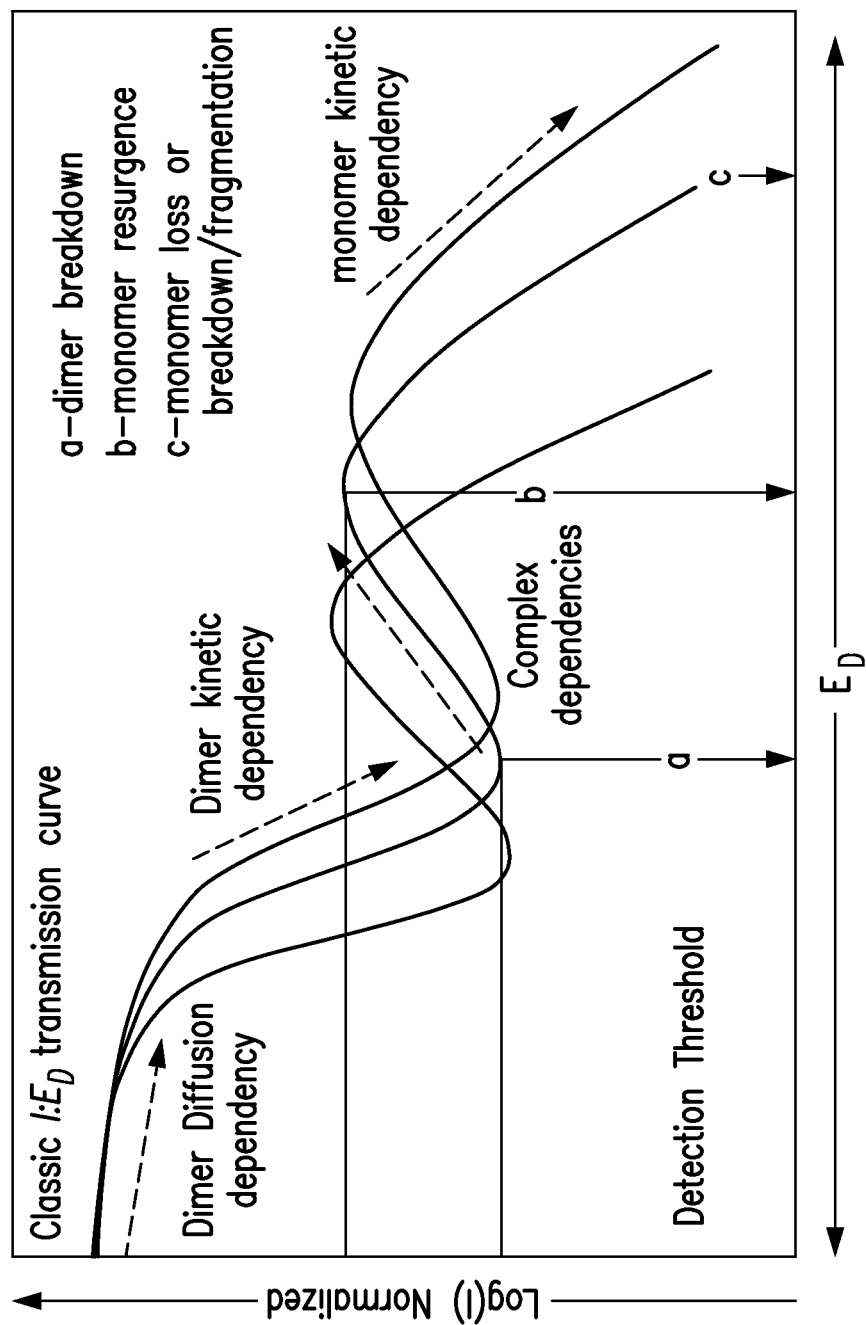


FIG. 3

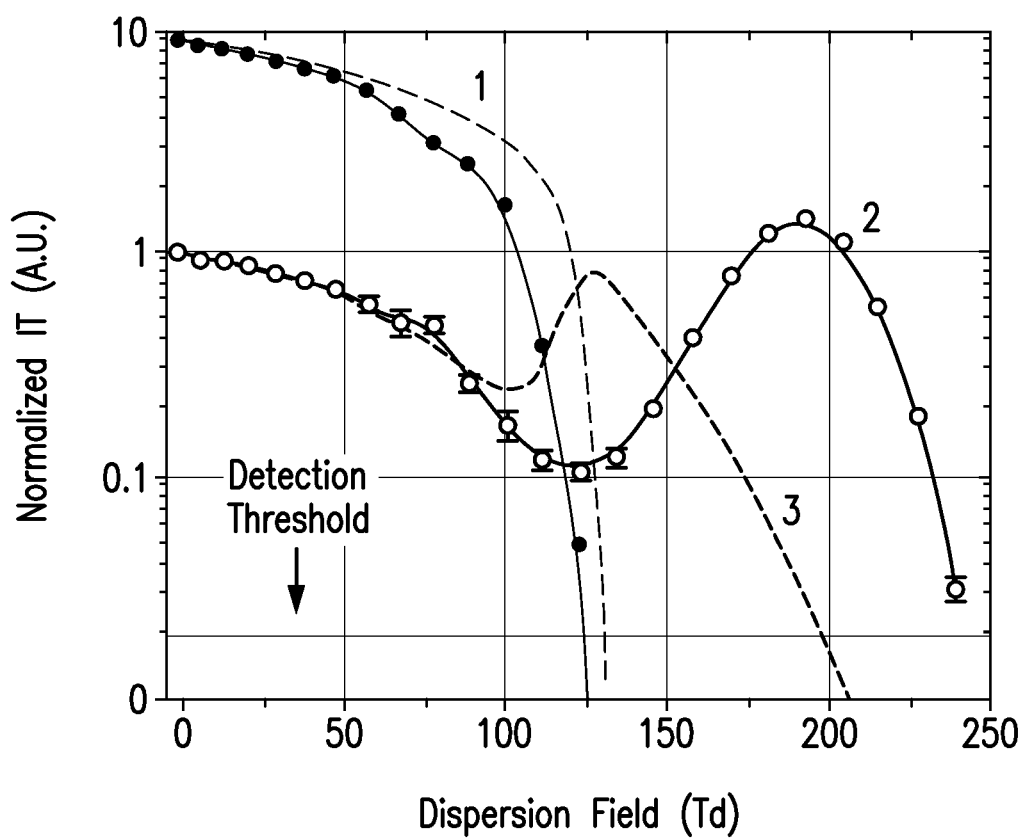


FIG. 4

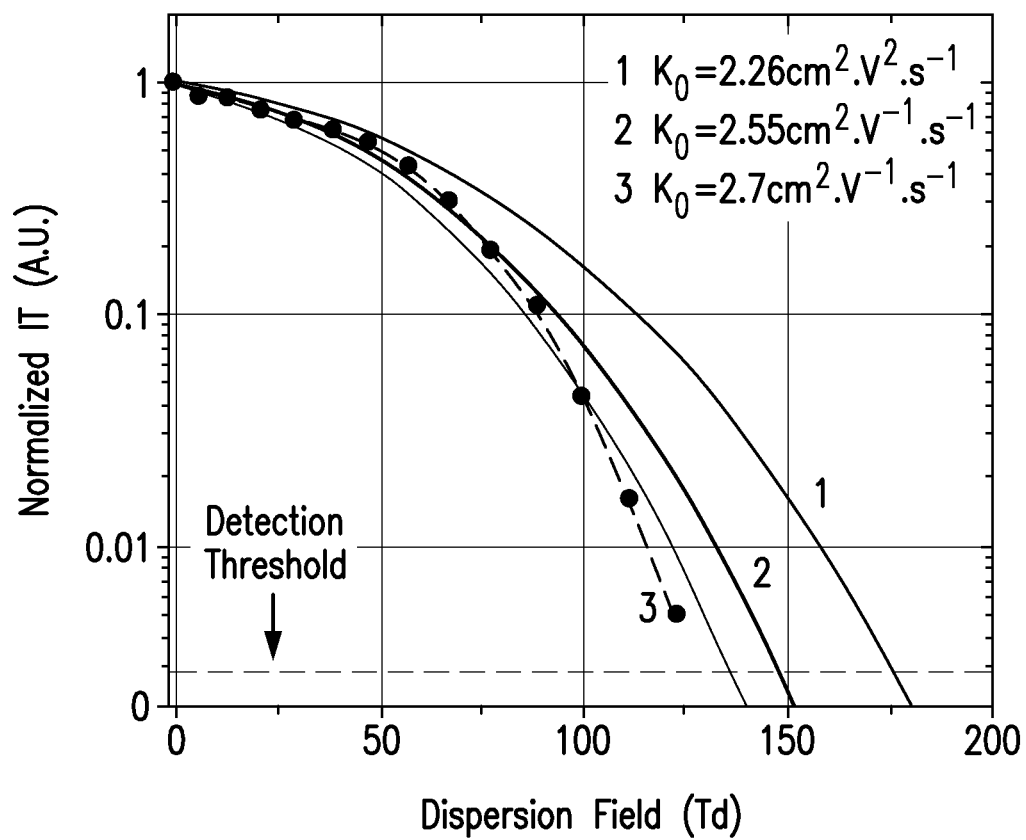


FIG. 5

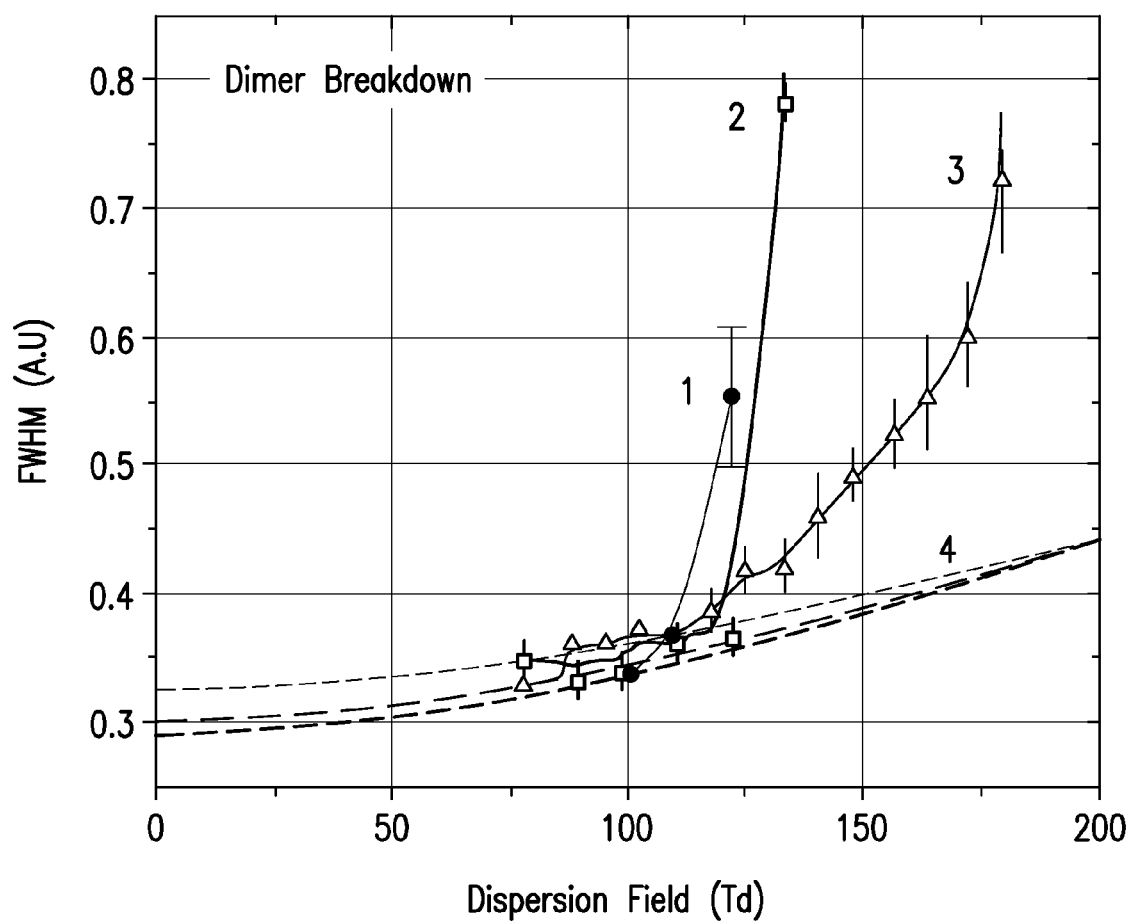


FIG. 6

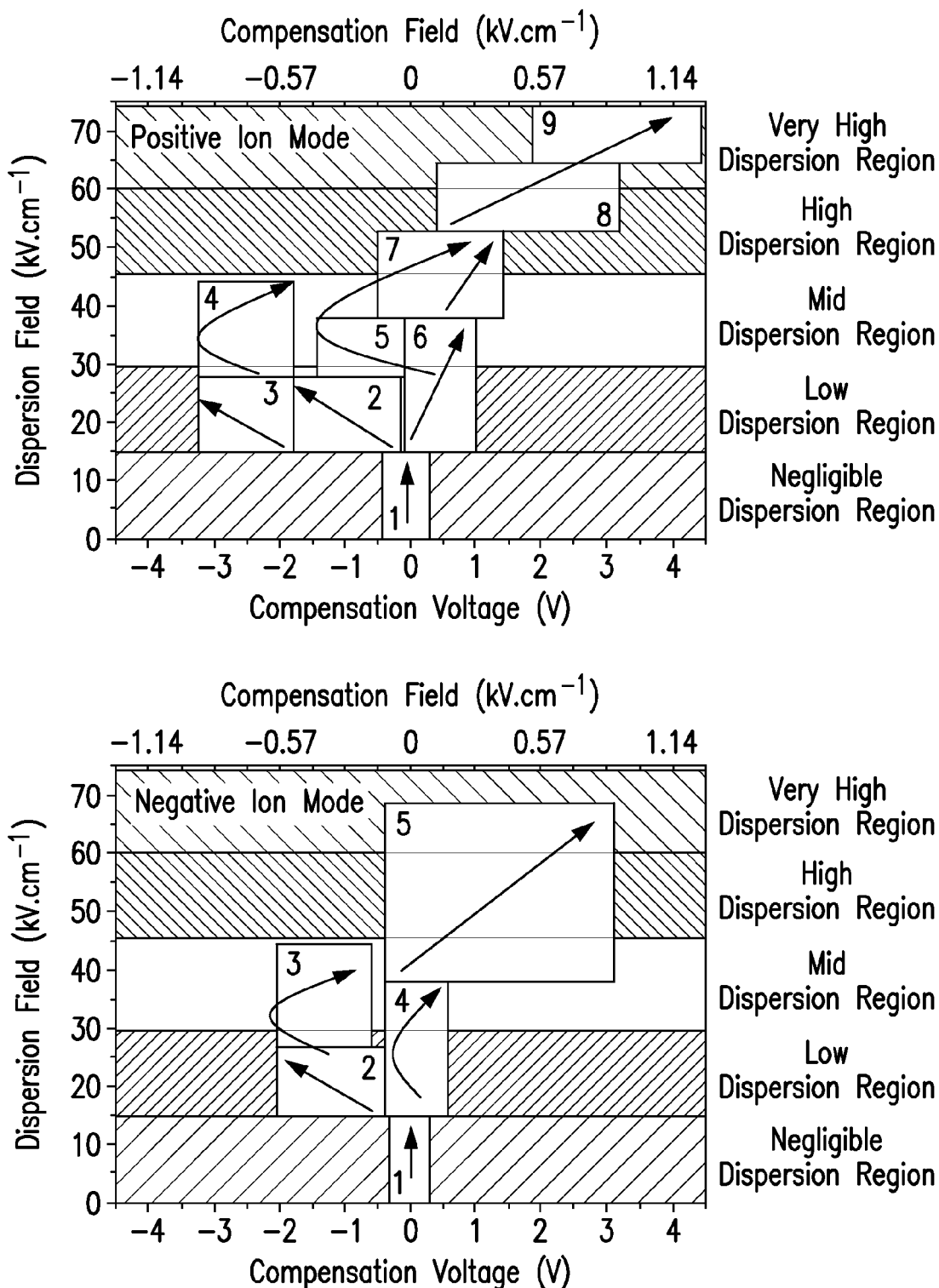


FIG. 7

1

FIELD ASYMMETRIC ION MOBILITY SPECTROMETRY SYSTEM

CROSS REFERENCE TO RELATED APPLICATIONS

This application claims priority to U.S. Patent Application Ser. No. 61/617,395 filed Mar. 29, 2012 which is incorporated herein by reference in its entirety.

FIELD OF THE INVENTION

The present invention relates to devices and methods for ion mobility systems. More specifically, the invention relates to a method and apparatus for detecting chemicals using a Field Asymmetric Ion Mobility Spectrometry (FAIMS) system.

BACKGROUND OF THE INVENTION

The ability to ionize gases using ion mobility systems is useful for a wide range of applications including many chemical detection applications. Ionization techniques, in which a gas sample is ionized and then separated into constituent parts that can be detected individually, are widely used for gas composition sensing. Two well-known examples are Ion Mobility Spectrometry (IMS) and Field Asymmetric Ion Mobility Spectrometry (FAIMS), also known as Differential Mobility Spectrometry (DMS). Ion mobility detection techniques tend to be very well suited to measuring trace constituents of gas mixtures that often consist of a carrier gas with additional gases mixed in at low concentrations (for example part-per-million or part-per-billion levels). Ion mobility techniques can also be used effectively over a range of gas pressures, including pressures close to one atmosphere. This makes them useful for, amongst other things, measuring low-level impurities in air. Because they work by measuring properties of ionized molecules and because gas samples for analysis generally consist mainly of neutral molecules, ion-mobility-based detectors generally incorporate an ionizer. The sample gas is passed through the ionizer to produce a population of ionized molecules that are then manipulated in some way involving separation or selection of ionized molecules according to their behavior in an electric field, before being detected. Ionizers commonly in use include radioactive sources, light-based devices such as ultra-violet lamps, and electrostatic devices such as corona discharge ionizers.

Practical chemical detectors such as Field Asymmetric Ion Mobility Spectrometry (FAIMS) systems compensate for varying environmental conditions as these can affect the output of the spectrometer. There are also a number of implementation challenges that the designer must overcome in order to produce a repeatable and reliable detection system. As one skilled in the art would recognize, a wide variety of non-idealities are encountered in practice. For instance, some of those most relevant to FAIMS systems fall into the following categories:

- Environmental and system variations and non-uniformities (e.g. changes in temperature, pressure, humidity and pump rate);

- Non-ideal device physics (e.g. charge mirroring); and

- Operational constraints (e.g. must provide certain levels of efficiency and stability while maintaining performance).

In this regard, it is to be understood and appreciated changes in ambient conditions such as temperature, pressure, and ambient humidity often lead to changes in FAIMS spectra. It is noted theoretical and/or empirical corrections exist

2

for these effects. While theoretical approaches work well for pressure and, to a lesser extent, temperature, not all environmental conditions can easily be corrected in this manner. Empirical corrections can be determined for all ambient variables, but this approach necessitates prior characterization of analytes across a multi-dimensional matrix of conditions and the storage of this data for each analyte, which can be a time consuming and laborious process. It is thus desirable to provide a system or approach that enables conditions at the sensor head to be controlled in such a system such that each analyte only has to be characterized under a small set of environmental conditions.

In regards to a FAIMS system, since it is a highly sensitive system, FAIMS spectrometers often detect down to parts-per-billion levels or more. Direct sampling from the ambient environment can thus be problematic for such highly sensitive systems, as they can be saturated by higher concentrations of analytes or interferents, thus hiding the analytes of actual interest. High concentrations of some analytes can persist within the system as contamination due to absorption onto internal surfaces, adversely affecting performance.

It is to be further understood that in Ion Mobility Spectrometry (IMS) systems (which work at a constant electric field) the peak location is used as an identifier and the parameter of Reduced Mobility (K_0) "factors" response variations due to temperature and pressure. Previous DMS/FAIMS systems (which work under variable electric field) have used the $E_C:E_D$ spectrum for identification but operate over a much limited field range ($35 \text{ kV}\cdot\text{cm}^{-1}$ c.f. $75 \text{ kV}\cdot\text{cm}^{-1}$).

SUMMARY OF THE INVENTION

The purpose and advantages of the invention will be set forth in and apparent from the description that follows. Additional advantages of the invention will be realized and attained by the devices, systems and methods particularly pointed out in the written description and claims hereof, as well as from the appended drawings.

One aspect of the invention relates to an apparatus, system and method for detecting, identifying, classifying and/or quantifying chemical species in a gas flow using a micro-fabricated ion filter coupled to a system adapted to apply drive signals to the ion filter. Coupled to the ion filter is a system adapted to measure the output of the ion filter, which in turn is coupled to a system adapted to extract numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow.

Other aspects of the invention is that inclusion of a FAIMS system in which the output of the FAIMS system is a two dimensional plot indicating the signal as a function of the RF voltage (Dispersion Field— E_D) and the compensation voltage (Compensation Field— E_C). This plot is used to discriminate between different chemicals. With reference to FIG. 1A, shown is an example $E_C:E_D$ matrix (Dispersion Spectrum) in which the spectrum is reduced to its Gaussian components (Peak Ion Current; Peak Width and Peak Position) and each discrete E_D point.

It is to be appreciated that another aspect of the present invention relates to finding information from plots characteristic of individual chemicals. An apparatus and method is provided to process variations induced by ambient conditions (specifically pressure, temperature and humidity). It is to be understood that when more than one species is present, chemical interactions and competitive ionization causes changes to both the spectra and signal strength. A noted advantage over the prior art is that the electric field strength

translating to the Dispersion Field (E_D) can be much higher, causing fragmentation of molecules. Fragmentation patterns provide characteristic, molecular specific classification features.

BRIEF DESCRIPTION OF THE DRAWINGS

The accompanying appendices and/or drawings illustrate various non-limiting, example, inventive aspects in accordance with the present disclosure:

FIG. 1A illustrates a Compensation Field-Dispersion Field matrix;

FIG. 1B illustrates a system overview of an illustrative embodiment of the invention;

FIG. 2 illustrates peak trajectories for Type A, B & C ions;

FIG. 3 illustrates integrated ion transmission Spectra;

FIG. 4 illustrates ion transmission spectra;

FIG. 5 illustrates transmission spectrum for non-fragmenting ions;

FIG. 6 illustrates peak width vs E_D for dimer ions; and

FIG. 7 illustrates positive and negative ion mode spectral character tables.

WRITTEN DESCRIPTION OF CERTAIN EMBODIMENTS OF THE INVENTION

The present invention is now described more fully with reference to the accompanying drawings, in which an illustrated embodiment of the present invention is shown wherein like reference numerals depict like elements. Before the present invention and specific exemplary embodiments of the invention are described, it is to be understood that this invention is not limited to particular embodiments described, as such may, of course, vary. It is also to be understood that the terminology used herein is for the purpose of describing particular embodiments only, and is not intended to be limiting, since the scope of the present invention will be limited only by the appended claims.

Where a range of values is provided, it is understood that each intervening value, to the tenth of the unit of the lower limit unless the context clearly dictates otherwise, between the upper and lower limit of that range and any other stated or intervening value in that stated range is encompassed within the invention. The upper and lower limits of these smaller ranges may independently be included in the smaller ranges is also encompassed within the invention, subject to any specifically excluded limit in the stated range. Where the stated range includes one or both of the limits, ranges excluding either both of those included limits are also included in the invention.

Unless defined otherwise, all technical and scientific terms used herein have the same meaning as commonly understood by one of ordinary skill in the art to which this invention belongs. Although any methods and materials similar or equivalent to those described herein can also be used in the practice or testing of the present invention, exemplary methods and materials are now described. All publications mentioned herein are incorporated herein by reference to disclose and describe the methods and/or materials in connection with which the publications are cited.

It must be noted that as used herein and in the appended claims, the singular forms "a", "an," and "the" include plural referents unless the context clearly dictates otherwise. Thus, for example, reference to "a stimulus" includes a plurality of such stimuli and reference to "the signal" includes reference to one or more signals and equivalents thereof known to those skilled in the art, and so forth.

With reference now to FIG. 1B, an illustrative embodiment of the invention is shown as a device for detecting, identifying, classifying and/or quantifying chemical species in a gas flow, depicted generally by reference numeral **10**. Device **10** includes a micro-fabricated ion filter **20** for separating ions in a gas flow. Preferably, filter **20** uses (but is not to be understood to be limited thereto) Field Asymmetric Waveform Ion Mobility Spectrometry (FAIMS) for separating ions. Examples of which can be found in commonly-assigned U.S. Pat. No. 7,498,570 entitled: ION MOBILITY SPECTROMETER; U.S. Pat. No. 7,714,277 entitled: SMART FAIMS SENSOR; U.S. Pat. No. 8,258,466 entitled: SMART FAIMS SENSOR; U.S. Pat. No. 8,227,749 entitled: PULSED FLOW ION MOBILITY SPECTROMETER; U.S. Patent Publication No. 2010/0148051 entitled: ION FILTER; PCT Publication Nos. PCT/2012/002843 entitled: SENSOR APPARATUS AND METHOD FOR USE WITH GAS IONIZATION SYSTEMS; PCT/2013/000404 entitled: FIELD ASYMMETRIC ION MOBILITY SPECTROMETRY FILTER; PCT/2012/065171 entitled: SENSOR APPARATUS AND METHOD FOR USE WITH GAS IONIZATION SYSTEMS; PCT/2013/020131 entitled: FIELD ASYMMETRIC ION MOBILITY SPECTROMETRY FILTER and PCT/2012/002856 entitled: CORONA IONIZATION DEVICE AND METHOD, each of which are hereby incorporated by reference in their entirety.

Device **10** further includes a drive signal system **30** for applying drive signals to the ion filter **20** and a measurement system **40** for measuring the output of the ion filter **20**. An extraction system **50** is coupled to measurement system **40** adapted and configured to extract numerical parameters from the measured output **40** of the ion filter **20**. A detection system **60** is preferably coupled to the extraction system **50** being adapted and configured to facilitate chemical detection, identification, classification and/or quantification of the gas flow of ion filter **20**.

As described further below, it is to be understood and appreciated that device **10** is adapted to use measurements of ion current as a function of one or more of compensation field and dispersion field to facilitate chemical detection, identification, classification and/or quantification which may preferably include the use of one or more ion peaks in FAIMS spectrum. It may further be adapted to classify ion peak trajectories into one or more of a number of categories and to use characteristic features of the trajectory of one or more ion peaks in the FAIMS spectrum to facilitate chemical detection, identification, classification and/or quantification wherein the characteristic features include ion peak position at one or more dispersion field values. Additionally, the characteristic features may be attributable to one or more of: ion fragmentation, dimer breakdown, monomer resurgence, monomer breakdown, field heating or changes in ion mobility. The characteristic features may include a point at which the gradient of compensation field versus dispersion field changes sign and/or a change in one or more peak characteristics at or in the vicinity of a particular dispersion field wherein the one or more peak characteristics include peak width and/or peak height. The one or more peak characteristics may further include peak location.

In additional embodiments of the invention, the rate of gas flow through the ion filter **20** is adjustable wherein the rate of gas flow is preferably selected to achieve a desired sensitivity and/or selectivity. Additionally, device **10** uses the relative peak amplitudes at a plurality of sample flow dilutions to infer one or more relative affinities wherein the one or more inferred relative affinities is preferably used to calculate one or more quantification correction factors. Still further, the one

5

or more inferred relative affinities is used in combination with one or more identified or partially identified chemical responses to calculate the affinity of unknown analytes to facilitate detection, identification, classification and/or quantification.

Additional embodiments of the invention are to be understood to include using measurements of ion current as a function of one or more of compensation field and dispersion field. Device **10** may monitor a vicinity of a reactant ion peak to detect and/or device **10** uses the presence of a dimer peak to facilitate chemical detection, identification, classification and/or quantification. Further, device **10** may use the dispersion field value at which a monomer peak undergoes one or more of; a characteristic shift in compensation field position, a split into two or more separate peaks to identify a point of monomer fragmentation. Device **10** may also be adapted and configured to use measurements of peak width along the dispersion field axis wherein the device **10** uses the dispersion field and compensation field co-ordinates of a point at which ion current drops below a threshold value to facilitate chemical detection, identification, classification and/or quantification.

Other embodiments of the invention may include using a character table and/or characteristic table to associate features of the spectrum with one or more classes of analytes to facilitate chemical detection, identification, classification and/or quantification wherein the features include one or more of dimer breakdown, monomer resurgence, a turning point in peak trajectory, monomer zero compensation field crossing point, and other FAIMS spectral features. Device **10** may be adapted and configured to compare analyte peak position to a position of the reactant ion peak to compensate for environmental effects wherein an analyte peak property is used in combination with a reactant ion peak property to ascertain identity, class and/or concentration of the analyte. Device **10** may further use an analyte peak property in conjunction with a reactant ion peak property to ascertain identity, class and/or concentration of the analyte.

It is also to be understood and appreciated the device **10** is operational such that the width of the ions species peak under zero dispersion field conditions is estimated by extrapolating from the measurements of the width under two or more non-zero field conditions wherein signal levels are normalized to the reactant ion peak intensity or total ion current and wherein one or more applied fields are adjusted to account for changes in number of molecules per unit volume of gas in proximity to the FAIMS ion filter **20**. Further, a number of molecules per unit volume are calculated by applying Van der Waal's equations for one or more gaseous components.

Device **10** may be further operational such that the relationship between peak parameters at a plurality of dispersion field frequencies is used to facilitate chemical detection, identification, classification and/or quantification wherein the region of the spectrum at which peak width starts to increase is also used to facilitate chemical identification, classification and/or quantification.

Device **10** may be adapted to adjust the frequency of the dispersion field waveform to modify the effects of humidity on resulting spectrum. The relationship between two peak intensities may be used to facilitate chemical detection, identification, classification and/or quantification. Device **10** may use the dispersion field value at which a dimer peak falls below a threshold value to identify a point of dimer breakdown and/or use an increase in a monomer peak intensity above a certain threshold dispersion field to confirm dimer breakdown to facilitate chemical detection, identification, classification and/or quantification wherein the relationship

6

between the two peak intensities is that of one peak increasing while the other decreases and wherein the dispersion field is adjusted to account for changes in one or more of pressure and the effective temperature of ions.

Device **10** may further detect and use changes in peak properties as a function of one or more of compensation field and dispersion field to infer a value for the mobility of an ion species, and/or device **10** may further detect and use the inferred mobility value associated with ion trajectory to create a temperature dependent correction factor that can be applied to the trajectory to facilitate chemical detection, identification, classification and/or quantification.

With the basic components of device **10** in accordance with an illustrated embodiment being described above, particular features of the present invention is further provided below.

In accordance with an embodiment of the invention, it is to be understood that by extracting the peaks (i.e. their Gaussian parameters) on each Compensation Field (E_C) line the path of a particular chemical is tracked. It is noted there are various different types of information that can be extracted from a chemical track, which are discussed below and can be used in singly or in combination depending on the degree of similarity of the chemicals under test.

With regards to the peak trajectories shown in FIG. **2**, it is to be understood both positive ions and negative ions derived from unknown chemicals samples and introduced to an analytical system can be placed into three broad categories (Type A, B, C) depending on the shape of their trajectory. It is noted this a broad and basic form of categorization. Specific, features of use, include (but are not limited to) those listed below. Track Inflection/Crossing Points/End Points

It is to be appreciated the Point of Inflection, Zero Crossing Point and End Point can be used to distinguish chemicals. The Point of Inflection is the point at which a curve turns back on itself. In the above example of FIG. **2**, the leftmost point of each blue line is the point of inflection for that curve. It is noted the point of inflection is a distinctive feature (most chemical trajectories have a point of inflection) notable exceptions are the RIP and some dimer peaks. The Zero Crossing Point is the point at which the peak trajectory crosses 0 CV. Hence, it is specific to Type B compounds. Note that under different conditions, such as differing humidity levels, a chemical can exhibit Type B or Type C behaviour. The End Point is the point at which the peak trajectory drops below a detection threshold for the peak.

Additionally, chemicals can be identified by matching the unknown track against previously collected tracks of known chemicals, which is termed Peak Location. It is to be understood the peak height (or area) plotted along the track of the peak contains distinctive features. Analytical systems in accordance with the present invention preferably operate at much higher fields (E_D) and the E_D dependant ion transmission contains much richer information than prior art systems (the kinetic and thermodynamic properties of ions are revealed and since these are molecular specific useful classification features are derived).

In accordance with the algorithm, the E_D/N scale may be converted to effective ion temperature and T_{eff} scales to $(E_D/N)^2$:

$$T_{eff} = T + \zeta \cdot M \cdot K_{(E/N)}^2 \cdot N_0^2 (E_D/N)^2 / (3k_b)$$

With reference to FIGS. **3** and **4**, shown is an Integrated Ion Transmission Spectra (FIG. **3**) and Resolved Ion Transmission spectra (FIG. **4**). This information is utilized to confirm the identity of a chemical or inferring information about its nature. Specifically the classification features of interest are:

- I. Dimer Formation—not all chemicals form dimers so this provides a degree of discrimination.
- II. Dimer Breakdown—the location and shape of the ion current curve at the dimer breakdown is indicative of the chemical(s) involved.
- III. Monomer Resurgence—at high fields dimers are unable to form and the signal from the monomer will increase.
- IV. Monomer Fragmentation—the monomer fragmentation occurs at high-field and as such provides information not available to prior art systems.

Additionally, physical constants for the ion generated from the chemical may be determined and include:

- V. Ion mobility (K_0)—(with reference to FIG. 5) Ion mobility (K_0) may be derived by a fit of empirical data to the following equation. K_0 is extracted from the g_{eff} term in the following fit to the data:

$$A_I = A_{I(in)} Q \exp(-t_{res} \pi^2 D_{II} / g_{eff}^2)$$

- VI. Collisional efficiency factor (Θ)— ζ may be determined from an analysis of the transmission profile and trajectory over a range of gas temperatures
- VIII. Peak width—it is noted the peak width plotted along the track can also contain distinctive information
- IX. Ion mobility (K_0)—Ion mobility at zero field is preferably defined by:

$$K_0 = \frac{16N \ln 2 \cdot k_b T}{N_0 \cdot q t_{res} \cdot W_{1/2}^2}$$

Wherein the zero field peak width ($W_{1/2}$) may be determined from an extrapolation of peak widths over a portion of the $E_C:E_D$ spectrum back to zero E_D .

- X. Ion breakdown field—it is noted the peak width turns exponential near the breakdown field. See FIG. 6 which illustrates Peak width vs E_D for dimer ions of Acetone, Butanone and DMMP wherein near dimer breakdown peak width turns exponential.

Additional features of the present invention include:

Environmental Correction/Compensation

It is to be appreciated that environmental conditions affect the track and the ion current (which is important for quantification). A noted tool for this purpose is the track called the RIP (Reactant Ion Products) which are the ions of nitrogen, oxygen and water that are ionised along with the sample of interest.

Normalize Ion Current Against RIP

As the quantity of ions is set by the ionisation method and the other ions that are presently working with ion currents normalised against the RIP will enable more accurate quan-

tification. In practice the chemical ion current is scaled against the height of the peak at low DF.

Normalize Location Against RIP

By measuring peak locations relative to the RIP it is possible to reduce the effect of environmental changes. This is likely to be most useful for humidity as temperature and pressure can be compensated for via physical models.

Compare RIP to Chemical Behavior

If there are chemicals with high electron/proton affinities present, the RIP will change characteristics enabling identification.

Effective Temperature and/or Pressure

It is noted that one method of finding chemical tracks that are invariant to ambient conditions is to use a y-axis scale that corresponds to the effective temperature and/or pressure. This includes a contribution from heating by the high field RF. Townsend units (i.e. field over number density— E/N), the effective ion temperature (T_{eff}) and the ion neutral collisional efficiency factors (ζ) may be used in the correction of temperature and/or pressure.

Humidity Banding

If compensation for humidity cannot be achieved then it is preferable to operate the device at a small number of fixed humidities which rules specific to each fixed humidity.

Flow Rate Changes

It is to be appreciated that flow rate changes allow improved sensitivity (important when working near the detection limit) and can also be used to confirm if a high affinity chemical is present.

Improved Townsend Correction

Townsend correction divides the electric field by the number density of the gas. Normally the calculation of the number density is performed by assuming that all the gases are ideal. In accordance with an embodiment of the invention, improved results are obtained by using van der Waal's equations for each component.

Higher Frequency Operation

It is noted that the impact of humidity on chemicals is thought to be due to the formation of clusters of water molecules around the chemical of interest in the low field and the breaking up of them in the high field. In accordance with an embodiment of the invention, by using higher frequencies the chemicals preferably remain clustered thus reducing the variability.

Using Characteristic Tables

It is to be appreciated that in the same way that Infra-Red is able to map specific frequencies to particular bonds, in accordance with an embodiment of the invention, particular tracks are associated with classes of molecules. This has been mentioned above regarding identifying chemicals as Type A, B or C. For instance, a non-exhaustive example of a UH-FAIMS character Table is shown below as:

Class	Analyte	Ion Mode	Chem. Functionality	Test Conc. Range	Gross Classification Rule
Nerve Agent	Tabun (GA)	Pos & Neg	Complex organo-phosphate	0.1-1.0 mg · m ⁻³	B, C
Nerve Agent	Sarin (GB)	Pos	Complex organo-phosphate	0.1-1.0 mg · m ⁻³	B, C
Nerve Agent	Soman (GD)	Pos	Complex organo-phosphate	0.1-1.0 mg · m ⁻³	B, C
Nerve Agent	VX	Pos & Neg	Complex organo-phosphate	<0.1	B, C

-continued

Class	Analyte	Ion Mode	Chem. Functionality	Test Conc. Range	Gross Classification Rule
Blister Agent	Sulfur Mustard (HD)	Neg	Reactive organo-chloro-sulfide	0.1-2.0 mg · m ⁻³	B, C
Blood/TIC	Hydrogen Cyanide (HCN)	Neg	Reactive LMW inorganic	12-90 mg · m ⁻³	A
Blood/TIC	Cyanogen Chloride (CNCl)	Neg	Reactive LMW inorganic	2.5-50 mg · m ⁻³	A (& B)
Nerve Simulant	Dimethyl methyl phosphonate (DMMP)	Pos	Stable organophos.	0.7-5.7 mg · m ⁻³	B, C
Blister Simulant	2-chloroethyl ethyl sulphide (2-CEES)	Pos	Stable organo-chloro-sulfide	0.1-2.0 mg · m ⁻³	B, C
Blister Simulant	Methyl Salicylate (MS)	Pos & Neg	Stable aromatic ester	0.1-2.0 mg · m ⁻³	B
Interferant (cleaning agent solvent)	Dipropylene Glycol Monomethyl Ether (DPGME)	Pos	Stable complex glycol	0.1-1.0 mg · m ⁻³	B, C
Interferant of ARDROX *	2,4 methyl-pentandiol	Pos	Diol	0.1-1.0 mg · m ⁻³	B, C
Interferant of floor stripper *	Monoethanol-amine	Pos	Hydroxylamine	Low < 1 mg · m ⁻³	B, C
Diesel Exhaust *	VOC/SO _x & NO _x mix	Pos & Neg	Hydrocarbon mix	—	A, C
Interferant of AFFF	Diethylene glycol mono butyl ether	Pos	Stable complex glycol	Low < 1 mg · m ⁻³	B, C
TIC	NH ₃	Pos	Reactive LMW inorganic	<1-100 ppm	A
TIC	Acrylonitrile (AcCN)	Neg	Reactive organo-cyanide	11-190 ppm	A, B
TIC	Hydrogen Sulfide (H ₂ S)	Neg	Stable LMW inorganic	10-138 ppm	B
TIC	Chlorine (Cl ₂)	Neg	Reactive LMW inorganic	7-50 ppm	A (& B)
TIC	Hydrogen Chloride (HCl)	Neg	Reactive LMW inorganic	7-50 ppm	A (& B)
TIC	Mono-methylamine	Pos	Amine	<0.1-5.0 ppm	B
VOC	Acetone	Pos	Ketone	0.2-2.1 mg · m ⁻³	B, C
VOC	Butanone	Pos	Ketone	0.3-3.2 ppm	B, C
VOC	1-Butanol	Pos	Alcohol	0.2-3.2 ppm	B, C
VOC	Iso-pentanol	Pos	Branched alcohol	Low ppm	B, C
VOC	Acetoin	Pos	Hydroxy ketone	Low ppm	B, C
VOC	Diacetyl	Pos	Diketone	Low ppm	B, C
VOC	Isoprene	Pos	Unsaturated hydrocarbon	Low ppm	B, C
VOC (nitro)	Nitromethane (CH ₃ NO ₂)	Pos	Stable Nitro	2.5-15 mg · m ⁻³	B
VOC (nitro)	Dinitrotoluene (DNT)	Pos & Neg	Stable aromatic Nitro	0.5-39 mg · m ⁻³	B, C

UH-FAIMS Character Tables

It is to be appreciated Character Tables are used in many forms of molecular spectroscopy to qualitatively define where different types of functional molecules give discrete responses in different parts of the available spectral window. They start off basic and become more detailed as more data is available, as illustrated in the positive and negative ion mode spectral character tables depicted in FIG. 7 and in the below UH-FAIMS character tables wherein table 1 depicts Discrete $E_C:E_D$ and response windows in the PIM and table 2 depicts discrete $E_C:E_D$ and response windows in the NIM.

TABLE 1

Win- dow	Key Information in this window - Peak Behavior (vs E_D)	e.g., Agents responding in this Window
1	Strong accentuation of Zero field Ion Current compared to absence of agent ITS is fairly flat in this region	All PIM responders
2	Ion transmission Turn in peak trajectory of low MW agents - resolution from RIP	Low MW agents and agents that form stable fragment ions (Acrylonitrile, nitromethane)
3	Sharp negative trajectory of Low MW agents Monomer turning point of intermediate MW agents	All PIM responders
4	All Dimer responses Dimers of smaller molecules begin to breakdown	PIM responders bar those that do not form stable dimers (Low MW agents)
5	Monomers of lower MW PIM responders	Low MW agents, Ion clusters of molecules that form stable fragment ions (Acrylonitrile, Nitromethane)
6	Proton bound dimers	Moderate MW PIM responders
7	Monomer Zero E_C crossing point of intermediate and higher MW agents Only stable dimers continue to yield a response but then breakdown Strong Positive E_C shift peak trajectories	Moderate MW PIM responders
8	Monomers of intermediate MW agents attenuated to below S:N or breakdown	VOCs
9	Only Monomers of higher MW agents respond Very strong positive E_C shift of monomer trajectories	Nerve agents DNT and other agents with ring structures

TABLE 2

Win- dow	Key Information in this window - Peak Behavior (vs E_D)	e.g., Agents responding in this Window
1	Accentuation or attenuation of Zero field Ion Current compared to absence of agent	Accentuation or attenuation of Zero field Ion Current compared to absence of agent
2	Ion transmission Turn in peak trajectory of low MW agents - resolution from RIP	Electronegative low MW TIC- Cl_2 , HCl, F_2 etc.
3	Low-mid MW PIM responders ion breakdown	SO_2 , AcCN, CH_3NO_2
4	Mid- MW NIM responders - high ion current, breakdown may occur	DMNB, DNT, MS
5	Mid-Higher NIM responders with ring structures	DNT

It is also to be appreciated that there may be chemicals which have very similar characteristics, in which case the

conditions can be changed (humidity, flow rate, temperature) to provide information to distinguish them.

The above presents a description of a best mode contemplated for carrying out the present invention, and of the manner and process of making and using them, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains to make and use these devices and methods. The present invention is, however, susceptible to modifications and alternative method steps from those discussed above that are fully equivalent.

Consequently, the present invention is not limited to the particular embodiments disclosed. On the contrary, the present invention encompasses all modifications and alternative constructions and methods coming within the spirit and scope of the present invention. Optional embodiments of the present invention may also be said to broadly consist in the parts, elements and features referred to or indicated herein, individually or collectively, in any or all combinations of two or more of the parts, elements or features, and wherein specific integers are mentioned herein which have known equivalents in the art to which the invention relates, such known equivalents are deemed to be incorporated herein as if individually set forth.

What is claimed is:

1. A method using a micro-fabricated ion filter for detecting, identifying, classifying and/or quantifying chemical species in one or more gas flows wherein one or more of the gas flows are adjustable, comprising:

applying drive signals to the ion filter;
measuring the output of the ion filter; and

extracting numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow wherein relative peak amplitudes at a plurality of sample flow dilutions are used to infer one or more relative affinities and wherein the one or more inferred relative affinities is used to calculate one or more quantification correction factors.

2. A method as recited in claim 1 wherein the rate of gas flow through the ion filter is adjustable.

3. A method as recited in claim 2 wherein the rate of gas flow is selected to achieve a desired sensitivity and/or selectivity.

4. A method as recited in claim 1 wherein the one or more inferred relative affinities is used in combination with one or more identified or partially identified chemical responses to calculate the affinity of unknown analytes to facilitate detection, identification, classification and/or quantification.

5. A method using an ion filter that separates ions using Field Asymmetric Waveform Ion Mobility Spectrometry (FAIMS) for detecting, identifying, classifying and/or quantifying chemical species in a gas flow comprising:

a micro-fabricated ion filter;

a system adapted to apply drive signals to the ion filter;

a system adapted to measure the output of the ion filter; and

a system adapted to extract numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow; and

using measurements of ion current as a function of one or more of compensation field and dispersion field to facilitate chemical detection, identification, classification and/or quantification wherein an analyte peak property is used in combination with a reactant ion peak property to ascertain identity, class and/or concentration of the analyte.

13

6. A method using an ion filter that separates ions using Field Asymmetric Waveform Ion Mobility Spectrometry (FAIMS) for detecting, identifying, classifying and/or quantifying chemical species in a gas flow comprising:

a micro-fabricated ion filter;
a system adapted to apply drive signals to the ion filter;
a system adapted to measure the output of the ion filter; and
a system adapted to extract numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow; and

using measurements of ion current as a function of one or more of compensation field and dispersion field to facilitate chemical detection, identification, classification and/or quantification wherein the region of the spectrum at which one or more peak parameters undergoes a change is used to facilitate chemical identification, classification and/or quantification.

7. A method as recited in claim 6 wherein the region of the spectrum used is a region at which the rate of peak width change with respect to dispersion field undergoes a change.

8. A method using an ion filter that separates ions using Field Asymmetric Waveform Ion Mobility Spectrometry (FAIMS) for detecting, identifying, classifying and/or quantifying chemical species in a gas flow comprising:

a micro-fabricated ion filter;
a system adapted to apply drive signals to the ion filter;
a system adapted to measure the output of the ion filter; and
a system adapted to extract numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow; and

using measurements of ion current as a function of one or more of compensation field and dispersion field to facilitate chemical detection, identification, classification and/or quantification wherein the frequency of the dispersion field waveform is selected and/or adjusted to modify the effects of humidity on a resulting spectrum.

9. A method using an ion filter that separates ions using Field Asymmetric Waveform Ion Mobility Spectrometry (FAIMS) for detecting, identifying, classifying and/or quantifying chemical species in a gas flow comprising:

a micro-fabricated ion filter;
a system adapted to apply drive signals to the ion filter;
a system adapted to measure the output of the ion filter; and
a system adapted to extract numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow; and

14

using measurements of ion current as a function of one or more of compensation field and dispersion field to facilitate chemical detection, identification, classification and/or quantification wherein the relationship between at least two peak intensities is used to facilitate chemical detection, identification, classification and/or quantification and further using the dispersion field value at which a dimer peak falls below a threshold value to identify a point of dimer breakdown to facilitate chemical detection, identification, classification and/or quantification.

10. A method as recited in claim 9 further including the step of using an increase in a monomer peak intensity above a certain threshold dispersion field to confirm dimer breakdown to facilitate chemical detection, identification, classification and/or quantification.

11. A method using an ion filter that separates ions using Field Asymmetric Waveform Ion Mobility Spectrometry (FAIMS) for detecting, identifying, classifying and/or quantifying chemical species in a gas flow comprising:

a micro-fabricated ion filter;
a system adapted to apply drive signals to the ion filter;
a system adapted to measure the output of the ion filter; and
a system adapted to extract numerical parameters from the measured output of the ion filter to facilitate chemical detection, identification, classification and/or quantification of the gas flow; and

using measurements of ion current as a function of one or more of compensation field and dispersion field to facilitate chemical detection, identification, classification and/or quantification and using changes in peak properties as a function of one or more of compensation field and dispersion field to infer a value for the mobility of an ion species to facilitate chemical detection, identification, classification and/or quantification.

12. A method as recited in claim 11 further including the step of using the inferred mobility value associated with ion trajectory to create a temperature dependent correction factor that can be applied to the trajectory to facilitate chemical detection, identification, classification and/or quantification.

13. A method as recited in claim 11 wherein the peak properties include one or more of peak height, peak width and peak area.

14. A method as recited in claim 13 wherein the peak properties are adjusted for changes in ion flux entering the ion filter.

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